# Determination of Acid Number and Fatty Free Acids (FFA) in Fats and Oils 

Titration Application
M101-01

## Introduction

The method is suitable for edible fats and oils such as butter, olive, palm or sunflower oil. The acid number is the quantity of base, expressed in milligrams of potassium hydroxide, that is required to neutralize all acidic constituents present in 1 g of sample. The calculation of the \% FFA depends on the titrated type of sample.

## Required Equipment

## Apparatus

$\qquad$

## Electrodes and Electrolyte

N 6480 eth pH electrode for titrations in non-aqueous mediaL 5034 electrolyte - ethanol with 1.5 mol// lithium chloride (LiCl/ethanol)
## Solutions

| $\square$ | Titrant: $\mathrm{KOH} 0.1 \mathrm{~mol} / \mathrm{L}$ in IPA (2-propanol). Also needed is $\mathrm{KOH} 0.1 \mathrm{~mol} / \mathrm{L}$ in ethanol. |
| :--- | :--- |
| $\square$ | Titer determination: Potassium hydrogen phthalate |Solvent: Ethanol/diethyl ether (1:1)

## Procedure

## Determination of the exact concentration of the KOH titrant

We recommend ready-to-use KOH titrants. The exact concentration of the $\mathrm{KOH} 0.1 \mathrm{~mol} / \mathrm{L}$ can be determined using the titrimetric standard potassium hydrogen phthalate.

In a 150 mL beaker, 0.2 g of the standard are weighed accurately and dissolved in 80 mL of distilled water with stirring. It is titrated with the $0.1 \mathrm{~mol} / \mathrm{L} \mathrm{KOH}$ solution.

Repeat the standardization two times. The average value is stored automatically in the exchangeable unit.


Figure 1: Titer

Page 1: Curve and result: Titer determination

## GLP documentation



## Calculation formula

| Titer: | $\left(W^{*} F 2\right) /\left((E Q 1-B)^{*} M^{*} F 1\right)->$ M103 |
| :--- | :--- |
| Mol (M): | 204.22000 |

Weight (W):
Blank value (B):
Statistics:
man
0.0000 ml Off

Factor 2 (F2):
Factor 1 (F1):
1000.0000 1.0000

## Method data overall view

| Method name: | Titer KOH | Created at: | 09/19/12 17:05:06 |
| :--- | :--- | :--- | :--- |
| Method type: | Automatic titration | Last modification: | 09/19/12 17:32:02 |
| Measured value: | mV | Damping settings: | None |
| Titration mode: | Dynamic | Documentation: | GLP |


| Measuring speed / drift: | Normal: | minimum holding time: 02 s <br> maximum holding time: 15 s <br> Measuring time: |
| :--- | :--- | :--- |
|  |  | 02 s |
|  |  |  |
| Drift: | $20 \mathrm{mV} / \mathrm{min}$ |  |
| Initial waiting time: | 0 s |  |
| Titration direction: | Decrease |  |
| Pretitration: | Off |  |
| End value: | Off |  |
| EQ: | On (1) | Value: |
| Slope value: | Steep |  |

Dosing parameter

| Dosing speed: | $100 \%$ | Filling speed: |
| :--- | :--- | :--- |
| Maximum dosing volume: | 50.00 ml |  |
| Unit values |  |  |
| Unit size: | 10 ml |  |
| Unit ID: | 00072696 |  |
| Reagent: | TBA Hydroxid |  |
| Batch ID: | 1.0265 |  |
| Concentration [mol/l]: | 0.10320 |  |
| Determined at: | $09 / 20 / 120: 57: 27$ |  |
| Expire date: | $04 / 12 / 12$ |  |
| Opened/compounded: | $10 / 19 / 11$ |  |
| Test according ISO $8655:$ | $12 / 01 / 10$ |  |
| Last modification: | $09 / 21 / 1215: 13: 56$ |  |
|  |  |  |

## Titration of the sample

Weigh the sample in a 100 mL beaker and add at least 50 mL of the solvent mixture to the sample. If necessary, heat the solution to dissolve the sample.

The sample weight should be calculated and selected that the titration amount is not more than 5 mL because of the long titration time.

For acid numbers between 0.2 and 1, the sample amount should be about $10-20 \mathrm{~g}$.
For acid numbers between 1 and 10, the sample amount should be about $1-3 \mathrm{~g}$.
Place the beaker on the magnetic stirrer and start the titration method. After the titration, rinse the electrode and burette tip with solvent. For each set of samples perform a blank titration with 50 mL of the titration solvent.

## Result Calculation

The enclosed titration example shows the calculation of the result in $\mathrm{mg} \mathrm{KOH} / \mathrm{g}$ sample (acid number).

The calculation of the \% FFA value depends on the titrated sample. For many oil and fat samples, the molecular weight of the oleic acid ( $282 \mathrm{~g} / \mathrm{mol}$ ) is used.
\% FFA $=(E Q 1-B) * 282 * T * 100 /(1000 * W)$
EQ1: mL consumption at the equivalence point
B: $\quad \mathrm{mL}$ consumption for the blank titration
282: molecular weight of oleic acid in $\mathrm{g} / \mathrm{mol}$
T: concentration of the KOH titrant (e.g. $0.1 \mathrm{~mol} / \mathrm{L}$ )
100: per 100 g sample
1000: conversation
W : sample weight in g

Blank titration page 1: Curve and result
GLP documentation


## Calculation formula

Blank:
EQ1 -> M02

Statistics
Off
Statistics: Off

Blank titration page 2: method

## Method data overall view

| Method name: | Blank AN |
| :--- | :--- |
| Method type: | Automatic titration |
| Measured value: | mV |
| Titration mode: | Linear |

## Measuring speed / drift: 12 s

| Initial waiting time: | 10 s |
| :--- | :--- |
| Titration direction: | Decrease |
| Pretitration: | off |
| End value: | off |
| EQ: | off |

## Dosing parameter

| Dosing speed: | $100.00 \%$ | Filling speed: |
| :--- | :--- | :--- |
| Maximum dosing volume: | 0.30 ml |  |
| Unit values |  |  |
| Unit size: | 10 ml |  |
| Unit ID: | 00072696 |  |
| Reagent: | TBA Hydroxid |  |
| Batch ID: | 1.0265 |  |
| Concentration [mol/l]: | 0.10350 |  |
| Determined at: | $09 / 21 / 12$ |  |
| Expire date: | $04 / 12 / 12$ |  |
| Opened/compounded: | $10 / 19 / 11$ |  |
| Test according ISO $8655:$ | $12 / 01 / 10$ |  |
| Last modification: | $09 / 21 / 1215: 28: 02$ |  |

> GLP documentation


## Method data

| Method name: | Acid number | Titration duration: | 3 m 33 s |
| :--- | :--- | :--- | :--- |
| End date: | 30.04 .13 | End time: | 12:19:19 |

## Titration data

Sample ID

EQ:
Olive oil
123.5 mV
$0.548 \mathrm{ml} /-47.4 \mathrm{mV}$
Weight:
End mV

AN mg KOH/g:
0.260

## Calculation formula

AN mg KOH/g:
$(E Q 1-B)^{*} T^{*} M^{*} F 1 /\left(W^{*} F 2\right)$

Blank value (B): $\quad 0.0990 \mathrm{ml}$ (M02)
Factor 1 (F1): $\quad 1.0000$
Factor 2 (F2): $\quad 1.0000$
$\mathrm{Mol}(\mathrm{M})$ :
$\begin{array}{ll}\text { Titre (T): } & 0.10350000(\mathrm{a}) \\ \text { Weight (W): } & 10.03650 \mathrm{~g} \mathrm{(m)} \\ \text { Statistics: } & \text { Off }\end{array}$

| Method data overall view |  |  |  |
| :---: | :---: | :---: | :---: |
| Method name: | Acid number | Created at: | 04/29/13 16 |
| Method type: | Automatic titration | Last modification: | 04/29/13 16 |
| Measured value: | mV | Damping settings: | strong |
| Titration mode: | Linear | Documentation: | GLP |
| Linear steps: | 0.050 ml |  |  |
| Measuring speed / drift: | User-defined: | minimum holding time: | 07 s |
|  |  | maximum holding time: | 20 s |
|  |  | Measuring time: | 04 s |
|  |  | Drift: | $10 \mathrm{mV} / \mathrm{min}$ |
| Initial waiting time: | 10 s |  |  |
| Titration direction: | Decrease |  |  |
| Pretitration: | Off |  |  |
| End value: | Off |  |  |
| EQ: | On (1) |  |  |
| Slope value: | Flat | Value: | 120 |

Dosing parameter

| Dosing speed: | $100.00 \%$ |
| :--- | :--- |
| Maximum dosing volume: | 6.00 ml |
| Unit values |  |
| Unit size: | 10 ml |
| Unit ID: | 00072696 |
| Reagent: | TBA Hydroxid |
| Batch ID: | 1.0265 |
| Concentration [mol/l]: | 0.10350 |
| Determined at: | $09 / 21 / 1222: 27: 50$ |
| Expire date: | $04 / 12 / 12$ |
| Opened/compounded: | $10 / 19 / 11$ |
| Test according ISO 8655: | $12 / 01 / 10$ |
| Last modification: | $09 / 21 / 12$ 15:28:02 |

## Contact Information

Please contact our titration experts if you have any application or product questions. Thanks!

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